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(21) International Application Number: PCT/US83/00913 (22) International Filing Date: 10 June 1983 (10.06.83) (71) Applicant: INTERNATIONAL COPPER ASSOCIATION, INC. [US/US]; 708 Third Avenue, New York, NY 10017 (US). (72) Inventors: SORENSON, John, R., J. ; 3415 West Mar- kham, Little Rock, AR 72205 (US). OBERLEY, Lar- ry, W. ; 151 Friendship Street, Iowa City, IA 52240 (US). (74) Agent: WEILD, David, III; Pennie & Edmonds, 330 Madison Avenue, New York, NY 10017 (US). (81) Designated States: AT (European patent), AU, BE (Eu- ropean patent), CH (European patent), DE (European patent), DK, FR (European patent), GB (European patent), JP, LU (European patent), NL (European pa- tent), SE (European patent).		Published <i>With international search report.</i>
(54) Title: COPPER COMPLEX FOR TREATING CANCER (57) Abstract Copper complexes exhibiting superoxide dismutase-like activity in mammalian cells. The copper complexes, which act as superoxide radical (O_2^-) scavengers, are used in treating cancer by replacing the lost superoxide dismutase activity which characterizes tumor systems. The low molecular weight and lipid solubility of the copper complexes facilitate penetration of cell membranes. Depending upon the specific type of copper complex used, treatment may result in decreased tumor growth, increased survival of the host organism, decreased tumor metastasis or induced morphological differentiation of cancerous cells. The copper complexes used according to the invention include copper salicylate complexes, their solvates as well as mixtures thereof. A method for the treatment of cancer using these copper complexes is also disclosed.		

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- 1 -

COPPER COMPLEX FOR TREATING CANCERTechnical Field

5 The present invention relates to a copper complex
for treating cancer by decreasing tumor growth, increasing
survival of a host, decreasing tumor metastasis and
inducing morphological differentiation of cancerous
cells. A method for treatment of cancer using this
10 complex is also disclosed.

Background of the Invention

 Oxygen toxicity in aerobic organisms has been
15 widely studied. The enzyme SOD - superoxide dismutase -
plays a significant role in the defense against such
toxicity.

 Superoxide radical (O_2^-) is generated in
20 various biological reactions; this free radical anion is
potentially damaging either because it is extremely
reactive or because it can generate highly toxic species
like OH. Superoxide dismutase catalyses the dismutation
of O_2^- :



 In a mammalian cell, two types of SOD are found.
One contains both copper and zinc and is located in the
30 cytosol and periplasmic space of the mitochondria
(CuZnSOD). The other enzyme contains manganese and is
present in the matrix of the mitochondria (MnSOD). All
normal mammalian cell types investigated contain these two
types of the enzyme, except erythrocytes which lack
35 MnSOD. However, in nearly all of the tumor systems



- 2 -

studied so far, the level of CuZnSOD and MnSOD activities were diminished. There are a few exceptions for CuZnSOD activity, but no exception has been found in the case of MnSOD activity. In many tumor samples, the MnSOD content is reduced to a level where it cannot be detected. Superoxide, on the other hand, is apparently still being generated in tumor mitochondria, and conceivably, the presence of this reactive species coupled with the relative lack of MnSOD could lead to changes in key subcellular structures due to the presence of oxygen derived chemical species. These changes may, in turn, contribute to the cancer phenotype.

The role of SOD in cancer has been the topic of a recent review in Oberley, L.W. and Buettner, G.R., The Role of Superoxide Dismutase in Cancer: A Review, Cancer Res. 39:1141-1149 (1979) and in 2 Oberley, L.W., Superoxide Dismutase, Chapter 6, (CRC Press 1982), both hereby incorporated by reference. SOD activities for numerous cancer types are specified.

Reduced MnSOD activity has been found in over fifty human, rat, mouse, spontaneous, transplanted virally-induced, chemically-induced, in vivo and in vitro tumors. In addition, both CuZnSOD and MnSOD activities in over thirty types of human tumors have been found to be less than those in control organs, Westman, N.G. and Marklund, S.L., Copper and Zinc-Containing Superoxide Dismutase and Manganese-Containing Superoxide Dismutase in Human Tissues and Human Malignant Tumors, Cancer Res. 41:2962 (1981), hereby incorporated by reference. Thus, it has been demonstrated that loss of SOD activity is characteristic of a wide variety of human tumors and is not a phenomenon restricted to mouse and rat tumors. Examples of cancer types characterized by low SOD



- 3 -

activity levels include: Morris hepatomas, H6 hepatoma, Novikoff hepatoma, C3H carcinoma, Lewis lung carcinoma, Walker carcinoma, Mammary adenocarcinoma, Ehrlich ascites, EL-4 ascites, Guerin T_R ascites, S91 melanoma, B16

5 melantic melanoma, Sarcoma 180, L1210 leukemia, L1210/6MP lymphoid leukemia, L1210/5FU lymphoid leukemia, P388 lymphoid leukemia, L1210/0 lymphoid leukemia, and Manning leukemia. This theory of excess free radicals with a deficiency of free radical scavengers in cancer cells has
10 also been proposed by Georgieff, K.K., Free Radical Inhibitory Effect of Some Anticancer Compounds, Science 173:537-539 (1971), hereby incorporated by reference. Georgieff showed that various types of chemotherapeutic
15 agents exhibit free radical inhibitory activity.

The present invention seeks to overcome the problems and disadvantages of the prior art. It has been discovered that copper complexes, which themselves act as
20 O₂⁻ scavengers, allow neoplastic cell to revert to its non-cancerous phenotype. These complexes also inhibit cancer cell growth in vivo, thereby prolonging the survival of tumor-bearing host organisms. Tumor metastasis is also reduced. The discovery is not only a
25 significant advance but also an unexpected discovery in the art of treating cancer.

Disclosure of Invention

Native SOD has a molecular weight of about 32,000
30 and, as a result, does not penetrate well into cell membranes. Cu(II) (3,5-diisopropylsalicylate)₂, CuDIPS, and other copper salicylate complexes are also O₂⁻ scavengers and have a molecular weight that facilitates penetration into cell membranes. Their smaller molecular
35 size (molecular weight between about 340 and about 1000,



preferably between about 340 and about 600 for copper salicylate complexes and molecular weight between about 845 and about 1800, preferably between about 845 and about 1400 for copper acetylsalicylate complexes), that is, low
5 molecular weight, and their lipid solubility allow better penetration of cell membranes, mitochondrial membranes and organelles. See Leuthauser, W.C., Oberley, L.W., Oberley, T.D., Sorenson, J.R.J., Ramakrishna, K., Antitumor Effect of a Copper Coordination Compound With Superoxide
10 Dismutase-Like Activity, JNCI 66:1077-1081 (1981), hereby incorporated by reference. Thus, CuDIPS can mimic the action of lost SOD activity in tumor cells. The use of copper salicylate complexes, their solvates, as well as mixtures thereof, for the treatment of cancer according to
15 this invention is advantageous in that the overall treatment mechanism does not, as with various known methods, involve only cytotoxicity but, instead, focuses on the replacement of lost SOD activity in tumor cells to effectively halt cell division and subsequent tumor
20 growth. Such a non-cytotoxic approach to cancer therapy allows a continued regimen of treatment which is not, as with methods using cytotoxic compounds, limited by potential compounding of toxic effects.

25 This invention involves the effect of compounds with SOD activity on tumor growth and survival of tumor bearing mammals. Natural SOD and the subject copper salicylate complexes were found to have surprisingly different effects on these parameters.

30 Low SOD activity, which is defined as SOD activity below that exhibited in a given normal, non-cancerous cell, characterizes various types of cancer. Copper salicylate complexes, their solvates, as
35 well as mixtures thereof, are useful in treating such types of cancer.



- 5 -

The purpose of tumor therapy is to eliminate the tumor or to retard its growth, prevent metastasis and at the same time, prolong survival of a host. Inhibition of tumor metastasis is an important parameter in cancer control. Most primary tumors can be easily treated by various therapy regimens, but metastasis is usually difficult to prevent.

The tumor model used in the following experiment is Ehrlich carcinoma cells, which metastasized to the lungs, forming macroscopic nodules. Microscopic metastases are observed as early as the 10th day after transplantation of tumor cells in Tween 80-saline treated control animals. When mice treated with 5 daily doses of CuDIPS were killed on the 45th day after transplantation, only microscopic nodules were observed in the lungs. CuDIPS had delayed the process of metastasis in these Ehrlich tumor mice. However, lung metastases were the cause of death of these treated mice. Prolonged treatment further increased survival, possibly by suppressing metastases longer and thereby retarding tumor growth.

The results of the experimentation of this invention indicate that CuDIPS does indeed exert in vivo antitumor activity in the solid form of Ehrlich carcinoma. A greater effect was observed with an increased number of treatments when the total dose did not exceed LD₅₀ of the compound. It has also been observed that in vitro, CuDIPS inhibits growth of cultured neoplastic cells and induces morphological differentiation in cultured neuroblastoma cells.

It has also been discovered that cancer types characterized by low SOD activity are advantageously treated using a copper salicylate complex in combination



- 6 -

with known anti-cancer drugs. For example, a treatment mixture containing both CuDIPS and bis-dichloroethyl-N-nitroso urea (BCNU) is effective in prolonging survival of tumor-bearing host organisms, 2 Oberley, L.W.,
5 (unpublished observation).

It has not been possible to discern the mechanism by which CuDIPS effects is antitumor activity. The mechanism is just not known. However, this mechanism is
10 not crucial to the advantageous practice of this invention. The following discussion merely postulates what are believed to be mechanisms that could describe the activity.

15 There are at least five different mechanisms by which CuDIPS could work: 1) cytotoxicity; 2) immune cell mediated effects; 3) antibody mediated effects; 4) inhibition of cell proliferation without cell
differentiation; 5) inhibition of cell proliferation with
20 differentiation. Cytotoxicity appears to be unimportant. Because tumors were not infiltrated with any type of inflammatory cell, the second mechanism also appears to be unimportant. Thus, either antibody mediated effects or
inhibition of cell proliferation with or without
25 differentiation appear to cause the following observed effects. These effects are unexpectedly opposite to the effects of other copper compounds that have been shown to treat cancer by killing tumor cells.

30 Best Mode for Carrying Out the Invention

Orgotein, a generic name for CuZnSOD derived from bovine liver, is obtained from Diagnostic Data
Incorporated, Mountain View, California. It is 98% pure
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- 7 -

CuZnSOD isolated from bovine liver and had 3,300 units of pure erythrocyte SOD activity/mg protein when assayed by the method of Beauchamp, C. and Fridovich, I., Superoxide Dismutase: Improved Assays and an Assay Applicable to
5 Acrylamide Gel., Anal. Biochem. 44:276-287 (1971), hereby incorporated by reference. Orgotein was dissolved in USP grade nonpyrogenic sterile saline.

CuDIPS is synthesized using known methods.
10 Sorenson, J.R.J., Copper Chelates as Possible Active Forms of the Antiarthritic Agents, J. Med. Chem. 19:135-148 (1976); Sorenson, J.R.J., Some Copper Coordination Compounds and their Antiinflammatory and Antiulcer
15 Activities, Inflammation 1:317-331 (1976), hereby incorporated by reference. CuDIPS is lipid soluble, and the LD₅₀ is 240 ± 33 mg/kg when administered subcutaneously to rats. CuDIPS is suspended in 10% Tween 80-saline ((Z) Sorbition mono-9-octadecanoate
20 poly(oxy-1,2-ethanediyl)) solution.

The SOD activity of Orgotein and of tumor homogenates is assayed by the method of Beauchamp and Fridovich. Reaction mixtures contain 0.05M potassium phosphate buffer (pH 7.8) with 10⁻³M
25 diethylenetriaminepentaacetic acid, 10⁻⁴M of xanthine, 5.6 x 10⁻⁵M of nitro blue tetrazolium, 10⁻³ units of xanthine oxidase and 1 unit of catalase. Addition of 5 x 10⁻³M NaCN to this mixture inhibits CuZnSOD activity and allows the measurement of the activity of the cyanide in-
30 sensitive MnSOD alone. The rate of blue formazan formation is measured by following the change of absorbance at 560 nm with a Cary Model 15 spectrophotometer. The sensitivity of the assay was such that 1 unit equaled 14 ± 1 ng of pure erythrocyte SOD.
35 The amount of protein in each sample is assayed using a



- 8 -

conventional method. Tissue homogenates are prepared by excising tumor mass from mice implanted with Ehrlich carcinoma cells. After bones and muscles are removed, the tumor tissue is rinsed and minced in 0.05 M potassium phosphate buffer (pH 7.8) and homogenized in a motor-driven Teflon pestle homogenizer. O_2^- scavenging ability of CuDIPS is determined by a similar assay method; the only difference is that dimethyl sulfoxide (DMSO) solvated potassium superoxide (KO_2) is used to generate O_2^- instead of xanthine and xanthine oxidase. CuDIPS is dissolved in 10% Tween 80-saline. This alternate method is chosen due to the length of previous procedure and lipid insolubility of some of the reactants of the previous assay. Free O_2^- is released when DMSO- KO_2 is added to an aqueous solution. Reaction mixtures contain 1:20 dilution of a saturated KO_2 solution (20 mg KO_2 /ml of dimethyl sulfoxide), $5.6 \times 10^{-5} M$ of nitro blue tetrazolium, 1 unit of catalase and 0-500 μg of CuDIPS in 0.5 M potassium phosphate buffer, pH 7.8. The formation of blue formazan is determined by recording the final absorbance of the assay mixture at 560 nm with a Cary Model 15 spectrophotometer. The sensitivity of this assay is found to be 0.85 μg of pure bovine SOD per unit of activity.

Tumor models used in these experiments are Sarcoma 180 cells implanted in 8-12 week old standard CF1 female mice (Charles River Breeding Laboratory, Wilmington, Massachusetts) and Ehrlich carcinoma cells implanted in standard CBA/J female mice (Jackson Laboratory, Bar Harbor, Maine). The tumor cells are maintained intraperitoneally in their respective hosts and are harvested and transferred weekly. Peritoneal fluid from these tumor hosts are centrifuged at 420xg for 2 minutes; the pellet is resuspended and washed three times



- 9 -

in isotonic saline. The resulting pellet is diluted 1:5 with isotonic saline. Cell counts are done using a Coulter Model B cell counter. Solid tumors are induced by implanting 5×10^6 cells intramuscularly into the right thigh of their respective hosts. Tumor growth is estimated by two dimensional measurements of the tumor thigh; group means and standard error of the means are plotted. The significance of differences between tumor growth curves is determined by a conventional Student's t-tests. Survival curves are analyzed by a conventional Mantel-Cox's Generalized Savage Test and a conventional Breslow's Generalized Wilcoxon test. The p values (probability that observation is due to chance alone) reported are the higher of the p values from these 2 tests.

The antitumor activity of Orgotein is studied in both tumor models. A dose of 1.2 mg of Orgotein in 0.15 ml Tween 80-saline is injected intramuscularly at the tumor site. A single dose is given to animals implanted with Sarcoma 180 cells at 1 hr, 5 days or 10 days after transplantation. Mice implanted with Ehrlich carcinoma cells are also treated with Orgotein. 5 or 10 daily doses are administered instead of a single dose; Tween 80-saline treated mice are used as control. Antitumor activity of CuDIPS is examined in mice with Ehrlich carcinoma only. A daily dose of 0.5 or 1.0 mg in 0.2 ml of Tween 80-saline is administered intramuscularly at the tumor site for 5 days, 10 days, or every Monday, Wednesday, and Friday for a total of 10 or 18 doses with the first dose administration at 1 hr following implantation of tumor cells. Another treatment schedule includes 5 daily doses beginning immediately after transplantation and an additional 5 daily doses starting on the 35th day after transplantation. Each treatment group consists of 24 tumor bearing mice. Tween 80-saline injected mice are used as controls.



- 10 -

Autopsies are performed on all dead animals. Tissues from the tumor masses and organs are excised and processed by known histologic techniques and stained with hematoxylin and eosin.

5

Antitumor Activity of Orgotein

Orgotein is administered to CF1 and CBA mice one hour after intramuscular implantation of Sarcoma 180 cells and Ehrlich carcinoma cells, respectively. As shown in Table I, tissue homogenates from these two types of solid tumor contain reduced amounts of total SOD activity and diminished amounts of MnSOD activity as compared to that from normal leg muscle of the mice without tumor cell implantation. Thus, these models appear to be representative in their SOD activity of the total class of tumors.

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- 11 -

TABLE I
SOD ACTIVITIES IN TISSUE HOMOGENATES (Units/mg protein)*

5	
	<div>Total SOD</div> <div>Mn SOD</div>
	<div>Normal leg muscle</div> <div>45 \pm 3 (4) **</div> <div>14 \pm 1</div>
	<div>Sarcoma 180</div> <div>39 \pm 6 (7)</div> <div>less than 2</div>
10	<div>Ehrlich carcinoma</div> <div>27 \pm 2 (13)</div> <div>less than 2</div>

* Sarcoma 180 and Ehrlich carcinoma tissues are excised between 5 and 45 days after transplantation, minced and homogenized in 0.05M potassium phosphate buffer (pH 7.8). Little differences in activity are observed between days 5 and 45, so the values obtained are averaged. Normal leg muscle tissue are treated in a similar fashion. SOD activity is measured by the conventional xanthine-xanthine oxidase nitro blue tetrazolium method. Errors represent standard error.

** Number of animals sampled.

Orgotein is assayed to have 49,500 \pm 300 units of SOD activity/mg protein by the xanthine-xanthine oxidase method. When 1.2 mg of Orgotein is given to mice with Sarcoma 180 tumors at 1 hour after tumor cell implantation, a reduction in tumor size is observed, as indicated in Chart I (p less than 0.05 between 11 and 27 days); after 27 days, the growth curves are the same.

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- 12 -

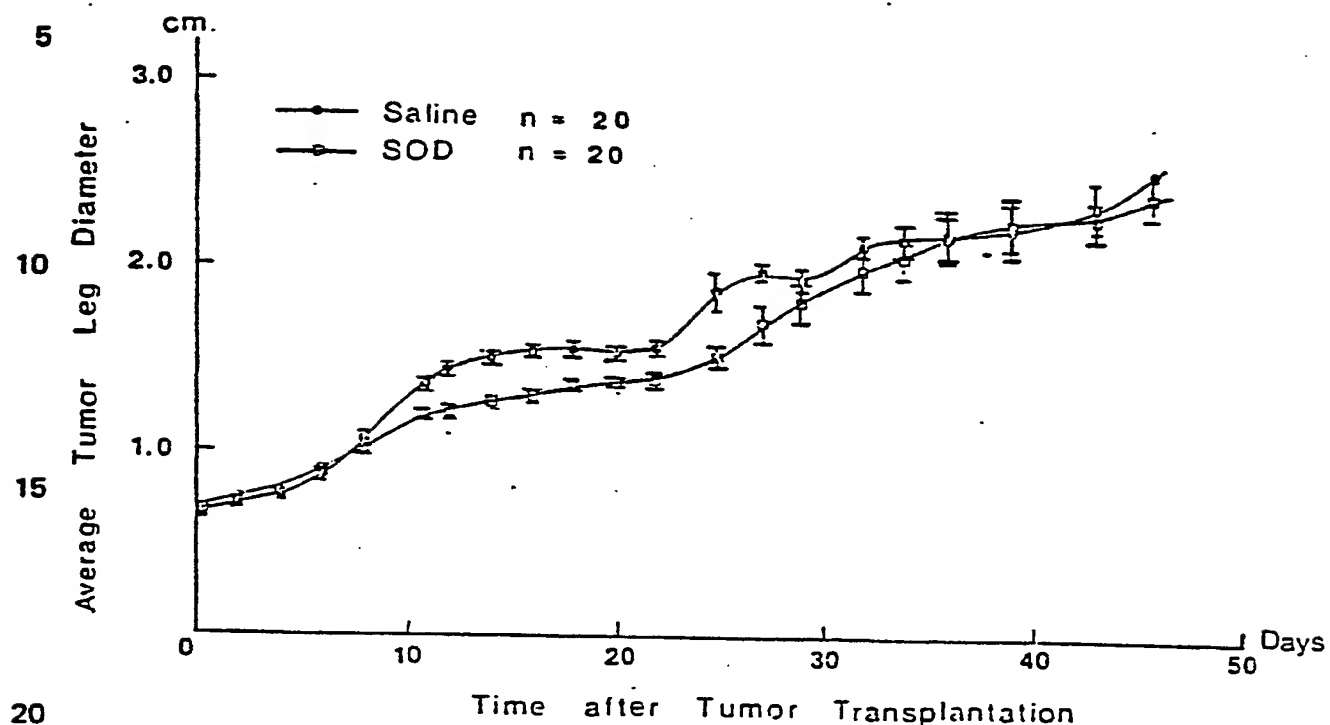
Chart I

Table II and Chart II show the survival times of these tumor bearing mice; there is a 23% increase in mean survival time in the Orgotein treated group (p less than 0.1). Viability of the Sarcoma 180 cells (2.5×10^7 cells/ml) is unchanged when they are incubated in 0.8 mg/ml Orgotein for one hour, showing that the effect of Orgotein is probably not due to cytotoxicity.

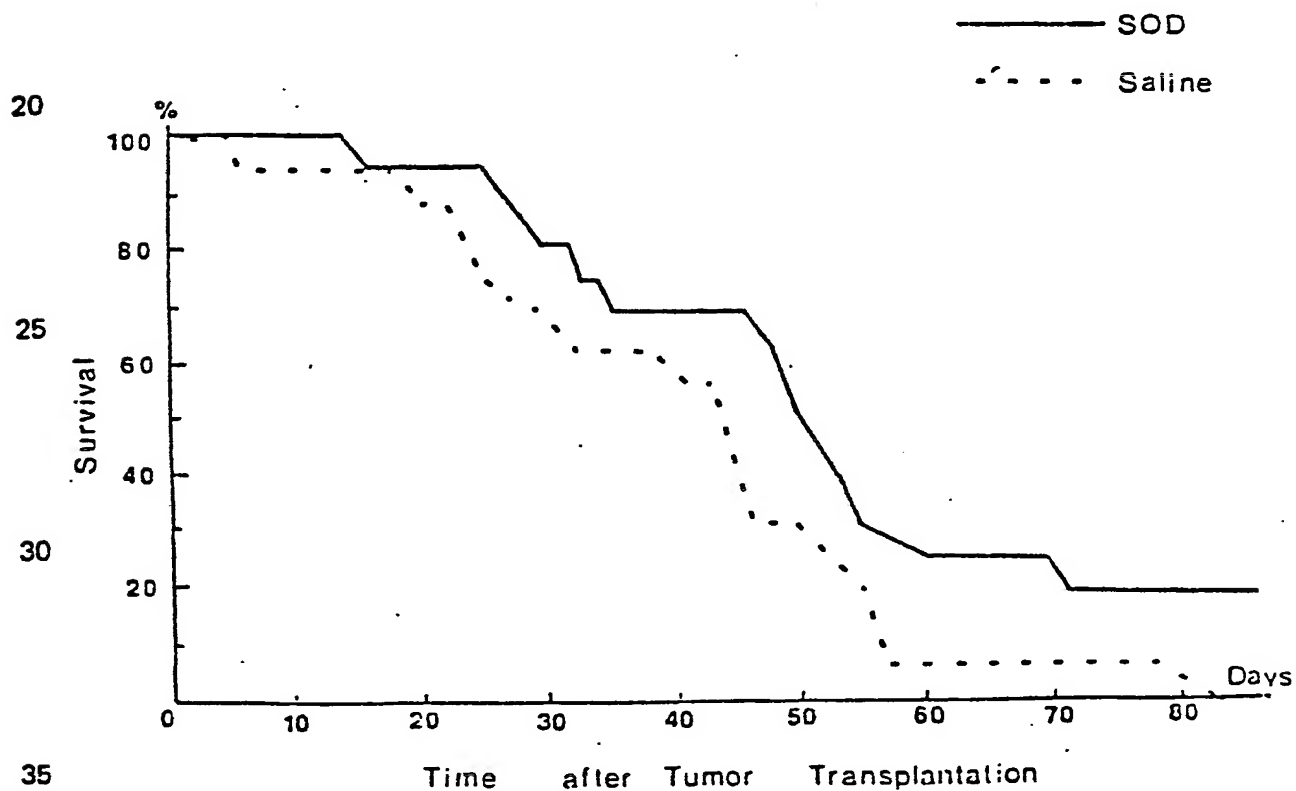
- 13 -

TABLE II
EFFECT OF ORGOTEIN TREATMENT ON SURVIVAL
OF CF1 MICE IMPLANTED WITH SARCOMA 180 CELLS

Treatment Group	Mean Survival Time in Days	Median Survival Time in Days	Maximum Survival Time in Days
Saline	41 \pm 5*	46	82
Orgotein	53 \pm 5	53	90

* standard error

Chart II



- 14 -

Tumors developed from those cells treated with Orgotein show no difference histologically from those developed from untreated cells. Histologic studies of the tumors from the Orgotein treated group show evidence of the development of a prominent fibroblast response in the area surrounding the tumor in some, but not all, of these animals on the 5th day after inoculation of tumor cells. An infiltrative pattern of growth is noted from tumors in the saline control mice.

10

When Orgotein at the same dose is given to mice transplanted 5 or 10 days earlier with Sarcoma 180 solid tumor, no significant decrease in tumor growth or increase in survival is observed compared to the saline treated control. Ehrlich carcinoma cells implanted in CBA mice also do not respond to Orgotein treatment administered on the day of implantation or for 5 or 10 consecutive days afterwards.

15

20 Antitumor Activity of CuDIPS

When SOD activity is measured by the method of DMSO- KO_2 , CuDIPS has a mean of $17,400 \pm 1,800$ units of SOD activity/mg of the compound, but the ligand DIPS alone, possesses no O_2^- scavenging ability. When Ehrlich carcinoma cells (25×10^7 cells/ml) are incubated with 5 mg/ml of CuDIPS for one hour, no toxicity is indicated. Preliminary dose study indicates that about 0.5 to 1.0 mg of CuDIPS gives a maximum effect without causing toxicity to animals. For this reason, a dose of 0.5 to 1.0 mg is always used. Chart III shows the difference in tumor growth when the tumor bearing mice are treated with CuDIPS (p less than 0.001) for 5 days.

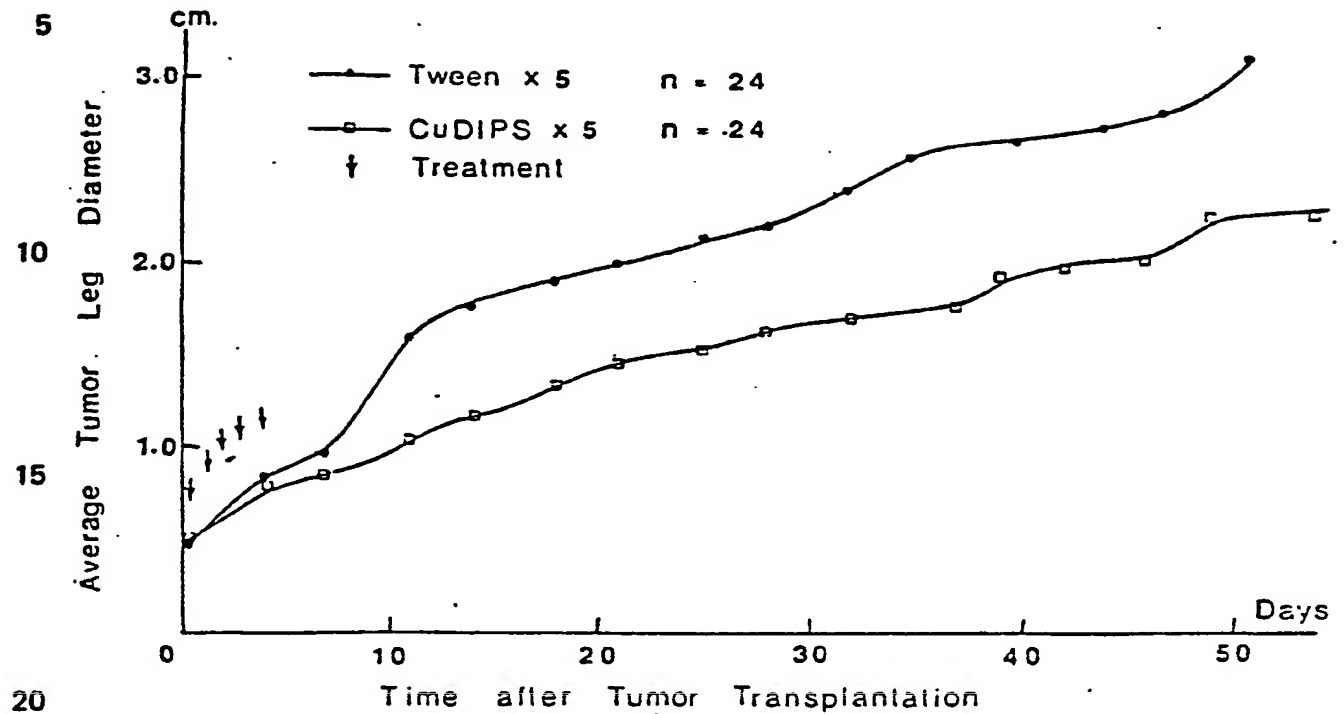
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- 15 -

Chart III

Approximately the same growth curve is obtained for all other treatment groups.

25 Table III summarizes the differences in survival times in the various treatment groups (p less than 0.0001 for all groups). Chart IV shows the survival curves of CuDIPS-treated groups.

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- 16 -

TABLE III
EFFECT OF CuDIPS TREATMENT ON SURVIVAL
OF CBA MICE IMPLANTED WITH EHRlich CARCINOMA CELLS

5	Treatment Group	Mean	Median	Maximum
		Survival Time in Days	Survival Time in Days	Survival Time in Days
10	Tween-80 x 5** (Daily)	33 \pm 4*	33	53
	CuDIPS x 5 (Daily)	54 \pm 3	52	68
	CuDIPS x 10 (Daily)	58 \pm 5	55	87
15	CuDIPS x 18 (MWF x 6)	50 \pm 4	49	79
	CuDIPS x 10 (MWF)	61 \pm 6	62	124
20	CuDIPS x 10 (Daily, days, 0-4 and 35-39)	58 \pm 4	59	greater than 124

* standard error

** total number of injections

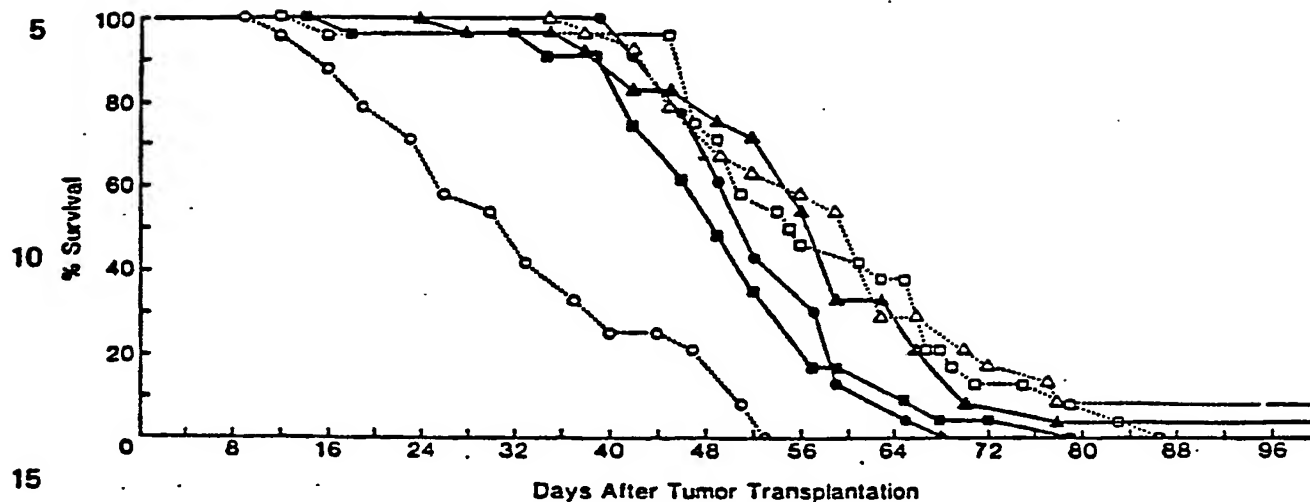
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- 17 -

Chart IV*

* Survival of CBA mice after implantations of Ehrlich carcinoma cells. Mice were given 5 doses of Tween 80 (O); 5 daily doses of CuDIPS (●); 10 daily doses of CuDIPS (□); CuDIPS on Mondays, Wednesdays, and Fridays for a total of 10 (Δ) or 18 (■) doses; or 10 daily doses of CuDIPS on days 0-4 and 35-39 (▲).

25

All treatments showed a similar effect on survival of the mice, but the group that received 0.5 mg of CuDIPS on Mondays, Wednesdays, and Fridays for a total of 10 doses showed a greater increase in mean survival. Out of 24 mice in this group, two of them had regressed tumors; however, the tumors recurred by 90 days after the initial implantation. Autopsies are performed on all dead mice. It is found that all animals died from massive lung metastases. Histologic studies of the organs of the tumor bearing mice revealed that microscopic nodules of tumor



- 18 -

cells are present in the lung of the control mice as early as 10 days after implantation of cells, but microscopic lung metastasis are not observed in the CuDIPS treated mice until 45 days after transplantation of the tumor cells. Histologic studies of the tumor from the CuDIPS treated mice show tumors are not infiltrated by any type of inflammatory cell.

CuSO₄ at a dose of 0.5 mg in 0.1 ml of Tween 80-saline is also given to CBA mice implanted with Ehrlich carcinoma cells for 5 days after transplantation. A continuous loss of weight of the tumor bearing mice is observed and death is caused by massive lung metastasis. In contrast to the above results, no weight loss was observed when mice are treated with various doses of CuDIPS, indicating that its effects are most likely not due to toxicity of the copper compounds.

Another group of CBA mice implanted with Ehrlich carcinoma cells received intramuscular injections of single doses of DIPS. No apparent SOD-like activity is exhibited by the DIPS and loss of body weight followed by death due to extensive lung metastasis is observed in the hosts.

CuSO₄ or DIPS alone have no antitumor activity. These observations demonstrate that the antitumor effects of CuDIPS is independent of and not attributable to any activity of its respective Cu or DIPS components.

The results indicate that the two tumor models chosen did indeed have a reduced level of both CuZnSOD and MnSOD activity. When this lost SOD activity is replaced by the local injection of Orgotein, a slight effect on the tumor growth and survival is observed in case of Sarcoma



- 19 -

180 tumor cells. Ehrlich carcinoma in CBA mice, however, did not respond to Orgotein treatment. Preliminary experiments with standard Swiss mice implanted with Ehrlich carcinoma cells indicated that a reduction in
5 tumor size occurs in the Orgotein treated mice. CBA mice are able to acquire immunologic tolerance; this may have contributed to the lack of response to the Orgotein treatment. Another possible explanation for the observed differences in response is that under certain conditions,
10 the tumor cells may be able to take in SOD by phagocytosis. Thus, a small amount of the enzyme may get in the cell or Orgotein may represent a source of copper complexes which can be utilized by cells. In any case, Orgotein does not have a large effect on the growth of
15 tumor cells in any of the models studied.

Other copper compounds have antineoplastic activity which prolongs survival of tumor bearing mice. Most of these compounds are cytotoxic; they prevent the
20 development of tumors by cell killing.

CuDIPS has been shown to be a potent anti-inflammatory agent. This compound is also non-toxic to the host at the dose levels used. The advantage of
25 this O_2^- scavenger over Orgotein is its small molecular size, which enables it to penetrate cell membranes and enter organelles, thus replacing the lost SOD activity within the tumor cells. The reduction in the size of the Ehrlich carcinomas in CBA mice after treatment
30 with CuDIPS means that CuDIPS has a marked beneficial effect on this tumor model. Because Orgotein exerts no effect on tumor growth in this model, and both compounds are O_2^- scavengers, these different antitumor activities may be due to differences in ability to
35 penetrate cell membranes. Thus, intracellular



- 20 -

localization of the added O_2^- scavengers may be responsible for the observed antitumor effect.

5 Due to the heterogeneous nature of tumors,
various types of tumors respond to the CuDIPS therapy
differently. In all CuDIPS treated tumor mice, tumor
growth retardation occurs; however, the reduction of tumor
size is unaffected by the total dose administered. This
10 could be due to the fact that the CuDIPS sensitive cell
populations are eliminated at a certain dose level;
different regimens may be required to achieve additional
beneficial therapeutic effects.

15 Growth Inhibition and Morphological
Differentiation Effects of CuDIPS On
Neuroblastoma Cells

20 CuDIPS inhibits cell population growth and
induces morphological differentiation of mouse
neuroblastoma cells in vitro.

The neuroblastoma cell clone NBP₂ provided by
Dr. K. N. Prasad (University of Colorado), is maintained
according to the procedure of Prasad and Hsie, Prasad,
25 K. N. and Hsie, A. N., Nature New Biol. 233:141 (1971),
hereby incorporated by reference. Cells are grown as a
monolayer in 75 cm² Falcon plastic tissue culture flasks
containing F-12 medium supplemented with 10% gamma
globulin-free new born calf serum, penicillin (100 U/ml)
30 and streptomycin (100 µg/ml). Cultures are maintained in
a humidified atmosphere of 5% CO₂ in a water jacketed
incubator at 37°C. The cells are harvested with 0.25%
Pancreatin and washed twice with fresh medium for
subsequent plating. Only exponentially growing cells are
35 used in the experiment. The viability of attached cells,



- 21-

determined 3 days after treatment by exclusion of trypan blue dye in control and treated cells, regularly exceeds 90%. Medium, serum, antibiotics, and pancreatin are purchased from Grand Island Biological Company.

5

CuDIPS was synthesized using known methods. Fifty-thousand cells were seeded in each 60 mm plastic petri dish containing 5 ml of the growth medium. Cells are treated with CuDIPS 24 hours after seeding. CuDIPS is dissolved in ethanol to give a final concentration of 10, 25, or 50 µg/ml of copper compound in the medium. The concentration of ethanol in the medium is kept constant at 1% in all the treated and control dishes. Culture dishes containing 1% ethanol serve as controls. One day after the start of treatments and daily afterwards, fresh drug and medium are provided in each dish. To avoid possible degradation, CuDIPS is dissolved in ethanol immediately before use. The experiment is done in triplicate. Three days after the initial treatments the number of cells and percent differentiation of cells are scored in each dish. Four hundred cells per culture dish from at least three different regions are examined with a Leitz Divert phase contrast microscope. Cells with one or more processes at least twice as long as the soma diameter (processes usually longer than 50 microns) are scored as differentiated. To count the number of cells, cells are harvested with Pancreatin and the number of cells in each dish is determined by counting cells in a hemocytometer.

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The effects on cell population and the percentage of morphological differentiation after three days of treatment with various concentrations of CuDIPS are observed and are summarized in Table IV.

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- 22 -

TABLE IV

Effects of Various Concentration of CuDIPS
on Cell Population and Differentiation of
Neuroblastoma in Culture

10	Treatments	Cell number* x 10 ⁵	% differentiation**
15	Control Cu(II) (3,5-diisopropyl- salicylate) ₂	17.1 ± 1.9	8.0 ± 3.7
20	10 µg/ml	12.7 ± 1.4	7.4 ± 2.6
	25 µg/ml	7.1 ± 1.2	22.4 ± 2.6
	50 µg/ml	1.4 ± 0.6	74.2 ± 4.9

* Each value represents mean of 24 determinations.

** Each value represents 9 determinations, at least
300 cells were counted in each dish.

The experiment is conducted in triplicate and each
value is given as mean ± S.D.

Three days after treatment with the CuDIPS at 10
µg/ml concentration, a small reduction in the number of
cells per culture dish is observed. The decrease in cell
population is significant (p less than 0.05, Student's
t-test). This level of concentration of CuDIPS caused no
significant change in the percentage of differentiated
cells. Treatment with 25 µg/ml of CuDIPS reduces the cell



- 23 -

population to $(7.1 \pm 1.2) \times 10^5$ cells/culture dish, a figure which represents less than half of the cell population found in control. At a concentration of 50 $\mu\text{g/ml}$, CuDIPS causes a remarkable reduction in cell population. Higher concentrations of CuDIPS (75 or 100 $\mu\text{g/ml}$) are toxic to cells and kills almost all cells in 3 days of treatments. No floating or dead cells are observed in the culture dishes which receive the copper compound to a maximum concentration of 50 $\mu\text{g/ml}$.

CuDIPS also induces morphological differentiation in neuroblastoma cells. Although the induction of differentiation is not observed at a concentration of 10 $\mu\text{g/ml}$, at a concentration level of 25 $\mu\text{g/ml}$, 22.4 ± 2.6 percent of the neuroblastoma cells differentiated. A treatment concentration of 50 $\mu\text{g/ml}$, produces an average of 74.2 ± 4.9 percent differentiated cells after 3 days of treatment. This represents more than 9 times the percentage of differentiated cells in controls. The majority of cells treated with 50 $\mu\text{g/ml}$ of CuDIPS morphologically resembles mature neural cells.

Antineoplastic Activities of Copper Salicylates

A variety of copper salicylate complexes exhibit antineoplastic activity in CBA/J female mice implanted with Ehrlich carcinoma cells. The degree of tumor growth inhibition positively correlated with the degree of lipid solubility of the particular copper salicylate complex.

Tumors are induced in inbred CBA/J female mice (The Jackson Laboratory, Bar Harbor, Maine) by implantation of Ehrlich carcinoma cells. Tumor cells are maintained intraperitoneally in their hosts and are



- 24 -

harvested and transferred weekly. Solid tumors are induced by intramuscular injections of 5×10^6 cells into the right thighs of hosts. Cells are enumerated with a Coulter Model B counter. Tumor growth is estimated by two-dimensional measurements of the thighs; group means and standard errors are calculated. The significance of differences between tumor growth curves is determined by Student's t-Tests for particular times. Survival curves are analyzed by Mantel-Cox's generalized Savage Test and Brelow's generalized Wilcoxon Test. The p-values reported are the higher values from these two tests. For survival data, p-values represent differences between complex treated and Tween 80-saline treated control groups over the whole curve and not at a particular time.

Copper salicylates are synthesized by known methods. Aqueous solutions of sodium salicylates are prepared by combining equivalent amounts of sodium hydroxide and the corresponding salicylic acid. To this solution is added 0.5 equivalents of copper chloride. The resulting precipitates are collected by filtration, washed with water, dried in vacuo for 24 hours at 50° , and submitted for elemental analysis. The analytical values agree with the calculated values within $\pm 0.4\%$.

Mice implanted with Ehrlich carcinoma cells are treated with copper chelates suspended in 10% Tween 80-saline solution just prior to administration. All treatments are delivered by 0.1 ml intramuscular injections at the site of cell implantation. A daily dose of 0.5 to 1.0 mg copper chelate in 0.1 ml of 10% Tween 80-saline is administered for 5 days, 10 days, or every Monday, Wednesday, and Friday for a total of up to 12 doses, with the first dose given 1 hour after inoculation of cells. Another schedule includes five daily doses



- 25 -

beginning 1 hour after transplantation. Each treatment group consists of 24 tumor-bearing mice. Mice given Tween 80-saline injections are used as controls.

5 The data presented in Charts V and VI show that each of the following copper compounds:

Cu(II) (salicylate)₂ - CuS

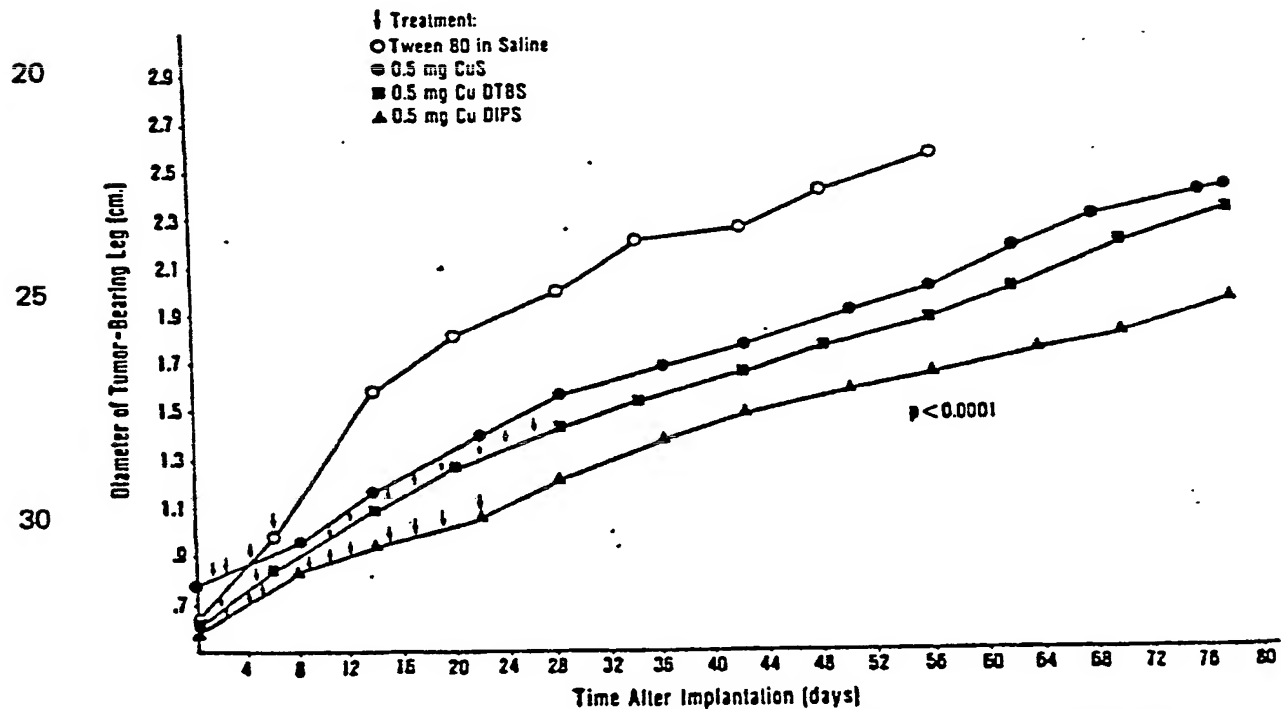
CuDIPS

10 Cu(II) (3,5-ditertiarybutylsalicylate)₂ - CuDTBS

dramatically reduced tumor growth and prolonged survival of the host organisms.

15

Chart V



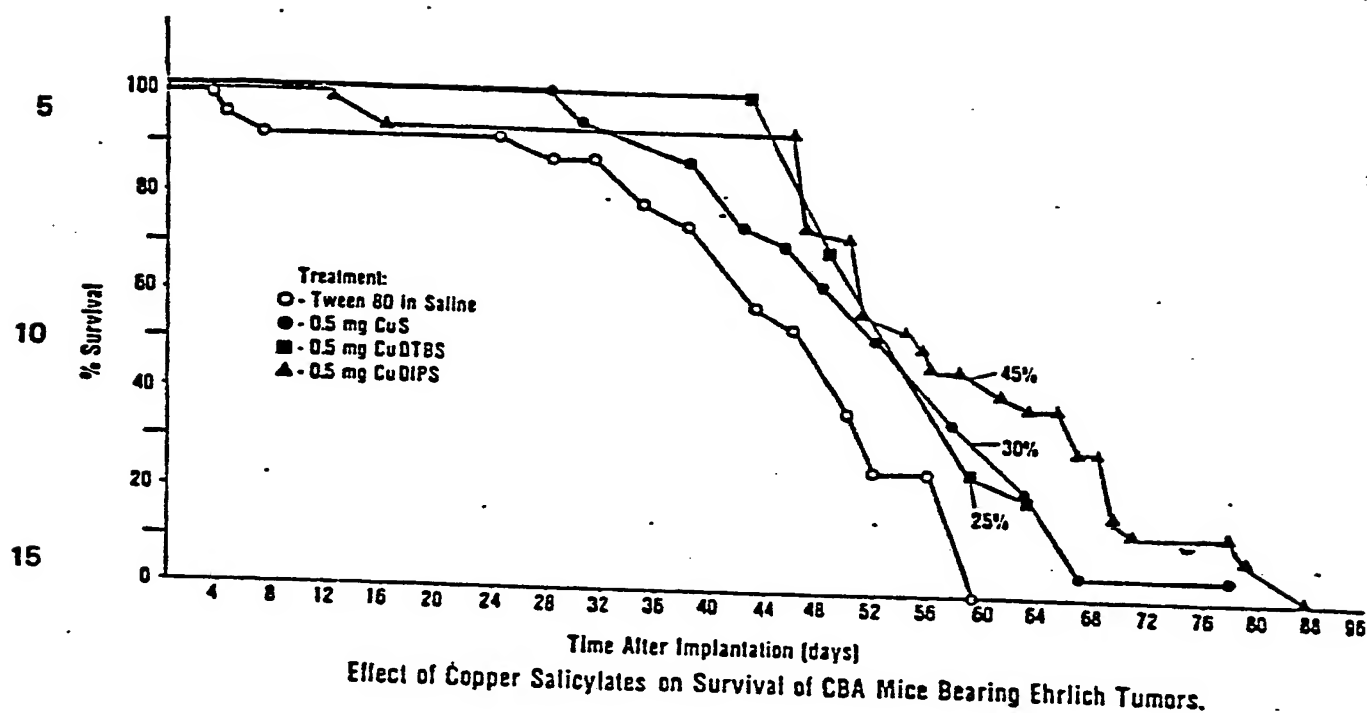
35

Effect of Copper Salicylates on Growth of the Ehrlich Tumor in CBA Mice.



- 26 -

Chart VI



On the 58th day after implantation, all of the control animals died, but there were surviving hosts in each of the complex-treated groups. 25% of the CuDTBS-treated, 30% of the CuS-treated, and 45% of the CuDIPS-treated mice survived beyond the time at which all of the control mice died. In addition, the period for 100% survival increased as compared to controls for all of the treated groups and the longest period was ten times longer, 44 days, for the CuDTBS-treated group. More than 90% of the animals treated with CuDIPS survived for 48 days. Lastly, when CuDIPS was given every day for 10 days, 10% of the animals survived for over 98 days, Leuthauser, S.W.C., Oberley, L.W. Oberley, T.D., Sorenson, J.R.J., and Ramakrishna, K., Antitumor Effect of a Copper Coordination Compound with Superoxide Dismutase-Like Activity, J. Nat'l. Cancer Inst. 66:1077-1081 (1981),



- 27 -

hereby incorporated by reference. Lung metastasis was markedly reduced in the copper complex treated mice.

5 The observed antineoplastic activity must be
attributed to the complexed form of copper because copper
sulfate and diisopropylsalicylic acid are shown to have no
antitumor effect in this Ehrlich tumor model, Oberley,
L.W., Leuthauser, S.W.C., Oberley, T.D. Sorenson, J.R.J.
and Pasternack, R.F., "Antitumor Activities of Compounds
10 with Superoxide Dismutase Activity" in Roles of Copper and
Other Essential Metalloelements in Inflammatory Diseases,
Humana Press, Clifton, New Jersey, pp. 423- 433 (1982),
hereby incorporated by reference.

15 Antitumor Activity of Copper
Acetylsalicylate Complexes

 The activity of various copper acetylsalicylate
compounds is also tested. As shown in Charts VII and
20 VIII, the copper complex of aspirin,
 $\text{Cu(II)}_2(\text{aspirinate})_4$, reduces tumor growth and
increases survival of the host. This complex is more
active than SOD, because the latter has no effect against
this solid Ehrlich tumor.

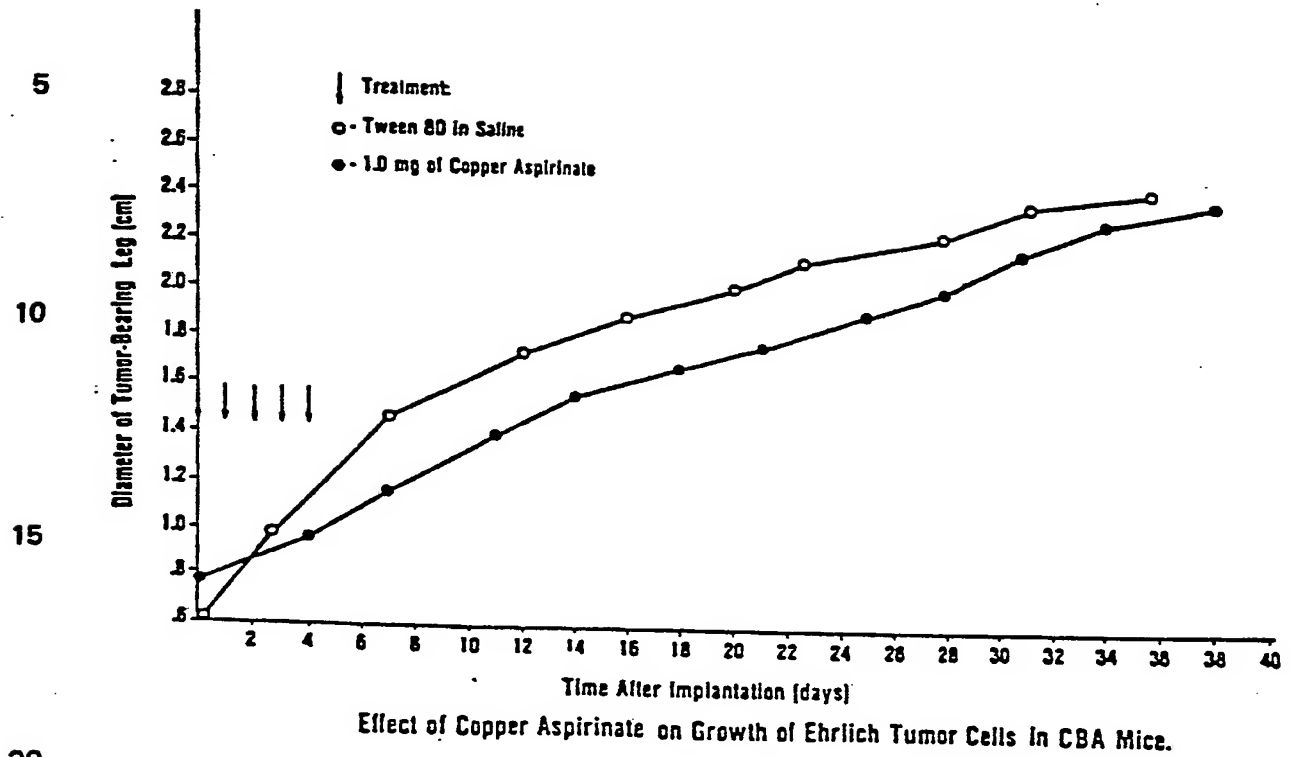
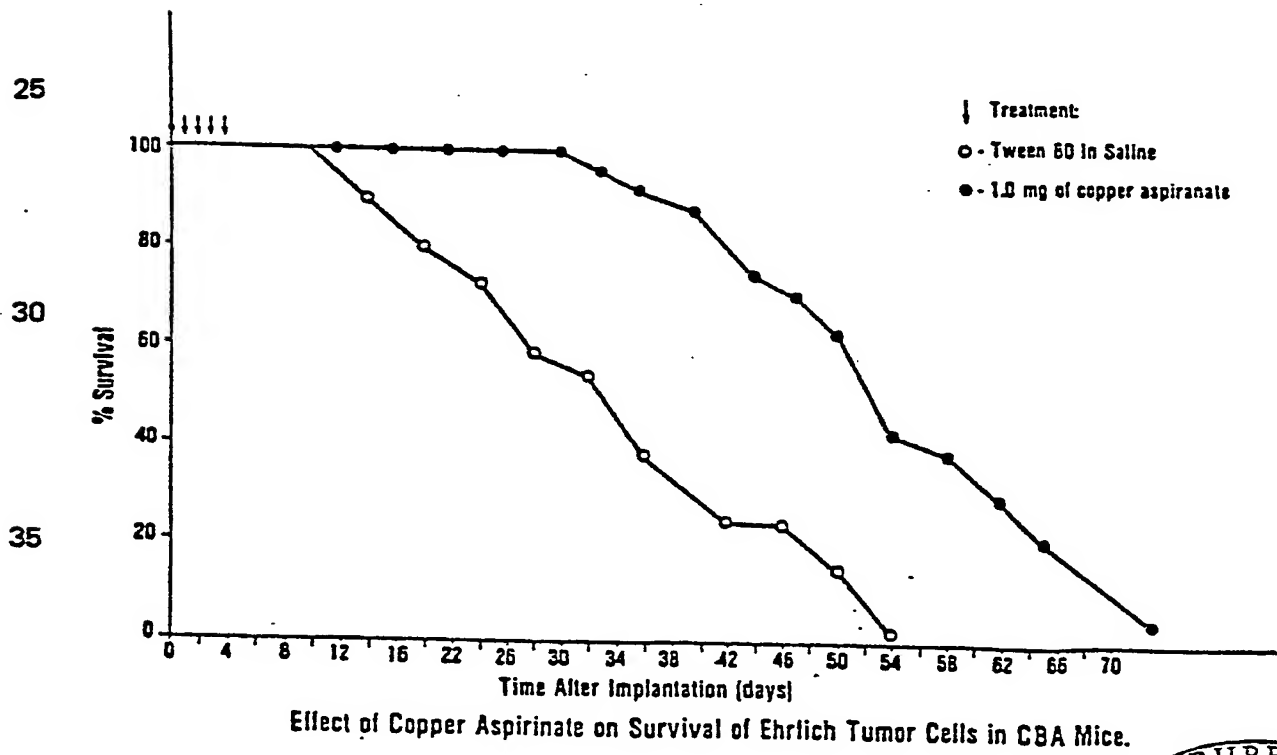
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- 28 -

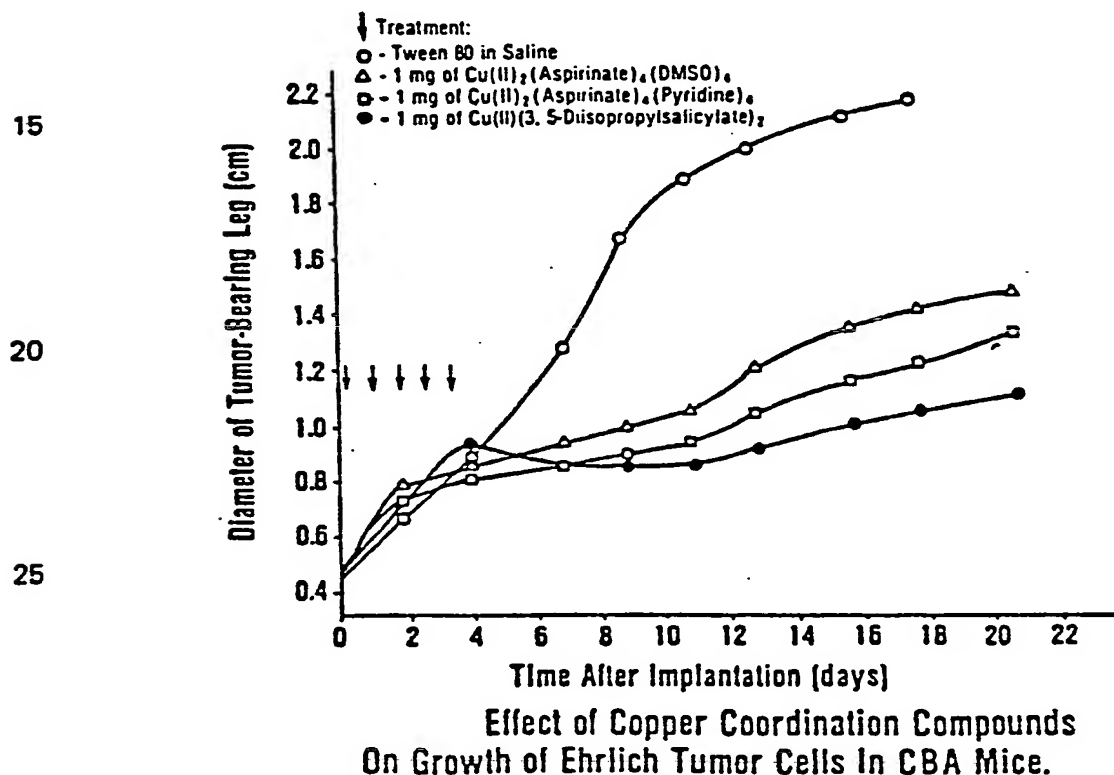
Chart VIIChart VIII

- 29 -

Tests involving other copper salicylate complexes reveal that they are more effective than the copper aspirinate compound, producing a further inhibition of tumor growth, as shown in Chart IX. The copper salicylate complexes compounds tested include:

$\text{Cu(II)}_2(\text{aspirinate})_4(\text{DMSO})_4$,
 $\text{Cu(II)}_2(\text{aspirinate})_4(\text{pyridine})_4$; and
 $\text{Cu(II)}(3,5\text{-diisopropylsalicylate})_2(\text{CuDIPS})$.

Chart IX



Among the copper complexes shown in Chart IX, increased lipid solubility appears to be related to increased antitumor activity. $\text{Cu(II)}(3,5\text{-diisopropylsalicylate})_2(\text{CuDIPS})$, having the highest degree of lipid



- 30 -

solubility of the tested compounds, is found to be most effective in reducing tumor growth.

5 Cytotoxicity does not appear to account for the
effectiveness of these complexes, because their LD₅₀
values are much larger than the dosages used in the
experiments. The effective dose of
Cu(II)₂(aspirinate)₄ is 50 mg/kg following
intramuscular administration, while its LD₅₀ is
10 determined to be 760 mg/kg in rats and greater than 320
mg/kg in mice, following subcutaneous injection. An
intramuscular dose of 25 mg/kg of CuS is found to be an
effective dose, while LD₅₀ values of 440 mg/kg, and
greater than 320 mg/kg were reported following
15 subcutaneous administration to mice, Sorenson, J.R.J.,
Rauls, D.O., Ramakrishna, K. Stull, R.E. and Voldeng, A.
N., "Anticonvulsant Activity of Some Copper Complexes" in
Trace Substances in Environmental Health, XIII University
of Missouri Press, Columbia, Mo, pp. 360-367 (1979).
20 Similarly, the effective intramuscular dose of CuDIPS is
25 mg/kg while reported LD₅₀ values are 240 mg/kg in
rats and greater than 320 mg/kg in mice following
subcutaneous administration. Also, if these complexes
cause cell death, a significant number of normal muscle
25 cells would have been killed with treatment. Finally,
there is no difference in the number of cells killed when
Ehrlich cells were incubated at 37°C for one hour with
medium containing 5 mg of CuDIPS per milliliter of saline.

30 In the treatment of human cancer patients, the
copper complexes would be administered to patients in
dosages and according to regimens that would be
therapeutically effective in replacing lost SOD activity
in the cancerous cells.

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- 31-

Having described the invention with particular
reference to the preferred form thereof, it will be
obvious to those skilled in the art to which the invention
pertains after understanding the invention, that various
5 changes and modifications may be made therein without
departing from the spirit and scope of the invention as
defined by the claims appended hereto.

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- 32 -

CLAIMSWe Claim:

- 5 1. A compound having superoxide dismutase-like activity comprising a copper salicylate complex, solvate or mixture thereof.
- 10 2. A compound for treating cancer in mammals comprising a copper salicylate complex, solvate or mixture thereof.
- 15 3. The compound according to claim 2 wherein the copper salicylate complex is $\text{Cu(II) (salicylate)}_2$.
4. The compound according to claim 2 wherein the copper salicylate complex is $\text{Cu(II) (3,5-diisopropyl-salicylate)}_2$.
- 20 5. The compound according to claim 2 wherein the copper salicylate complex is $\text{Cu(II) (3,5-ditertiary-butylsalicylate)}_2$.
- 25 6. The compound according to claim 2 wherein the copper salicylate complex is $\text{Cu(II)}_2(\text{aspirinate})_4$.
- 30 7. The compound according to claim 2 wherein the solvate of a copper salicylate complex is $\text{Cu(II)}_2(\text{aspirinate})_4(\text{pyridine})_4$.
8. The compound according to claim 2 wherein the solvate of a copper salicylate complex is $\text{Cu(II)}_2(\text{aspirinate})_4(\text{DMSO})_4$.
- 35 9. The compound according to claim 2 wherein the compound has a molecular weight which enables the compound



- 33 -

to penetrate cell membranes, mitochondrial membranes and organelles.

5 10. The compound according to claim 2 wherein the compound has a molecular weight between about 340 and about 1000.

10 11. The compound according to claim 6 wherein the compound has a molecular weight of between about 845 and about 1800.

 12. The compound according to claim 2 wherein the compound is soluble in lipids.

15 13. The compound according to claim 2 wherein the compound is an O_2^- scavenger.

20 14. A compound for treating cancer types in mammals which are characterized by low levels of superoxide dismutase activity comprising a copper salicylate complex, solvate or mixture thereof.

25 15. The compound according to claim 14 for treating Ehrlich carcinoma.

 16. The compound according to one of claims 14 or 15 wherein the copper salicylate complex is $Cu(II)(salicylate)_2$, $Cu(II)(3,5-diisopropylsalicylate)_2$, $Cu(II)(3,5-ditertiarybutylsalicylate)_2$ or $Cu(II)_2(aspirinate)_4$.

35 17. The compound according to one of claims 14 or 15 wherein the solvate of a copper salicylate complex is $Cu(II)_2(aspirinate)_4(pyridine)_4$ or $Cu(II)_2(aspirinate)_4(DMSO)_4$.



- 34 -

18. The compound according to claim 14 wherein the compound has a molecular weight between about 340 and about 1000.

5 19. The compound according to claim 14 wherein the compound has a molecular weight of between about 845 and about 1800.

10 20. A compound for treating cancer types in mammals which are characterized by low levels of superoxide dismutase activity comprising a copper salicylate complex and an anti-cancer drug.

15 21. The compound according to claim 20 wherein the copper salicylate complex is $\text{Cu(II)(salicylate)}_2$ and the anti-cancer drug is bis-dichloroethyl-N-nitroso urea.

20 22. A method for treating cancer in mammals comprising administration of a therapeutically effective amount of a copper salicylate complex, solvate or mixture thereof.

25 23. The method according to claim 22 for treating Ehrlich carcinoma.

30 24. The method according to claim 22 wherein the copper salicylate complex is $\text{Cu(II)(salicylate)}_2$, $\text{Cu(II)(3,5-diisopropyl-salicylate)}_2$, $\text{Cu(II)(3,5-ditertiarybutylsalicylate)}_2$, or $\text{Cu(II)}_2(\text{aspirinate})_4$.

35 25. The method according to claim 22 wherein the solvate of a copper salicylate complex is $\text{Cu(II)}_2(\text{aspirinate})_4(\text{pyridine})_4$ or $\text{Cu(II)}_2(\text{aspirinate})_4(\text{DMSO})_4$.



- 35 -

26. The method according to claim 22 for treating cancer types in mammals which are characterized by low superoxide dismutase activity.

5 27. A method for treating cancer types in mammals which are characterized by low superoxide dismutase activity comprising administration of a therapeutically effective amount of a mixture of a copper salicylate complex and an anti-cancer drug.

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28. The method according to claim 27 wherein the copper salicylate complex is $\text{Cu(II)(salicylate)}_2$ and the anti-cancer drug is bis-dichloroethyl-N-nitroso urea.

15 29. The method according to claim 22 including decreasing tumor growth.

30. The method according to claim 22 including decreasing tumor metastasis.

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31. The method according to claim 22 including inducing morphological differentiation of neuroblastoma cells.

25 32. The method according to claim 22 including inhibiting growth of neoplastic cells.

33. The method according to claim 22 including prolonging survival of the mammal.

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INTERNATIONAL SEARCH REPORT

International Application No PCT/US83/00913

I. CLASSIFICATION OF SUBJECT MATTER (If several classification symbols apply, indicate all) *

According to International Patent Classification (IPC) or to both National Classification and IPC:
C07F 1/08; A01N 37/36; A61K 31/60; A01N 55/02; A61K 31/30

II. FIELDS SEARCHED

Minimum Documentation Searched *

Classification System

Classification Symbols

U.S.

424/230; 424/294; 260/438.1

Documentation Searched other than Minimum Documentation
to the extent that such Documents are included in the Fields Searched *

DARC/QUESTEL: CAS ONLINE: DIALOG

III. DOCUMENTS CONSIDERED TO BE RELEVANT ¹⁴

Category *	Citation of Document, ¹⁵ with indication, where appropriate, of the relevant passages ¹⁷	Relevant to Claim No. ¹⁸
X	US, 4,221,785, published 9 September 1980 Sorenson	1-7 & 9-19
X	US, 4,287,190, published 1 September 1981 Boettcher et al.	1-3
X	Chemical Abstracts, Vol. 85, issued 1976 (Columbus, Ohio, USA) Sorenson et al., G.D. Searle and Co., Chicago, Ill. "Some Copper Coordination Compounds and Their Antiinflammatory and Antiulcer Activities", See page 40, Column 2, Abstract No. 116777t, Inflammation (N.Y.) 1976, 1(3), 317-31 (Eng).	6
Y	Chemical Abstracts, Vol. 69, issued 1968 (Columbus, Ohio, USA), Kratsmar et al. (Pharm. Fak., Komenski-Univ., Bratislava, Czech.) "Copper (II) Complexes with Organic Ligands. VIII Copper (II) Salicylate and 0-cresotate complexes with ethanol", see page 10573, Column 1 - Column 2, Abstract No. 113013z, Chem. Zvesti, 1968 22(7), 493-501(Ger)	1, 2, 7 & 9-19

* Special categories of cited documents: ¹⁵

- "A" document defining the general state of the art which is not considered to be of particular relevance
- "E" earlier document but published on or after the international filing date
- "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
- "O" document referring to an oral disclosure, use, exhibition or other means
- "P" document published prior to the international filing date but later than the priority date claimed

"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step

"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.

"Δ" document member of the same patent family

IV. CERTIFICATION

Date of the Actual Completion of the International Search ¹

13 September 1983

International Searching Authority ¹

ISA/US

Date of Mailing of this International Search Report ²

26 SEP 1983

Signature of Authorized Officer ²⁰

Helen M. S. Sneed

III. DOCUMENTS CONSIDERED TO BE RELEVANT (CONTINUED FROM THE SECOND SHEET)

Category *	Citation of Document, ¹⁶ with indication, where appropriate, of the relevant passages ¹⁷	Relevant to Claim No ¹⁸
Y	Chemical Abstracts, Vol. 84, issued 1976 (Columbus, Ohio, USA) Sorenson, (G.D. Searle and Co., Chicago, Ill). "Copper Chelates as active forms of the antiarthritic agents", see page 14, Column 2, Abstract No. 38591r, J. Med. Chem. 1976, 19(1), 135-48 (Eng).	1, 2, 6, 9-13, 14-16 18, 19
Y	US, 4,287,190, published 1 September 1981 Boettcher	8
Y	US, 3,884,949, published 20 May 1975 Eicke et al.	1-5, 9-16 18, 19
A	US, 2,480,084, published 23 August 1949 Meyer	20-33
A	US, 4,373,953, published 15 February 1983 Deinet et al.	20-33
A	US, 4,291,062, published 22 September 1981 Leigh et al.	20-33

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